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AlPO 4: 1.000 mol, or 26.98 g Al; Al 2 Cl 6: 1.994 mol, or 53.74 g Al; Al 2 S 3: 3.00 mol, or 80.94 g Al; The Al 2 S 3 sample thus contains the greatest mass of Al. 27 . 3.113 × × 10 25 C atoms

Answer Key Chapter 3 - Chemistry 2e | OpenStax

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Chapter 3 ChemiCal Compounds 69 ook around you. Do you think you see anything composed of just one element... any objects consisting only of carbon, or of gold, or of hydrogen? The correct answer is almost certainly no. If you are lucky enough to have a diamond ring, you have a piece of carbon that is almost pure (although a gemologist would tell you

Chapter 3 - An Introduction to Chemistry: Chemical Compounds

Chemistry 101 ANSWER KEY REVIEW QUESTIONS Chapter 3 1. What mass of chlorine is present in 12.2 g of PbCl2? 12.2 g PbCl 2 x 1 mol 2 Cl 35.45 g x x = 3.11 g 278.1 g 1 PbCl 2 1 mol 2. How many atoms of oxygen are present in 2.15 g of Ca3(PO4)2?

Chemistry 101 ANSWER KEY 1 REVIEW QUESTIONS Chapter 3 1 ...

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